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### **(54) Emollient durability enhancing siloxanes.**

(57) A method of enhancing the durability of an emollient material on a substrate by forming a mixture of an emollient material and an effective amount of an organosilicon compound, and applying the mixture to the substrate to be treated. The organosilicon compound is either an aminofunctional, amidofunctional, or carboxyfunctional, polysiloxane. A skin conditioning composition of enhanced durability is also disclosed which is a mixture of an emollient material and an effective amount of an organosilicon compound which is either an aminofunctional, amidofunctional, or carboxyfunctional, polysiloxane.

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This invention relates to the enhancement of the durability of emollient materials with certain organosilicon compounds. More particularly, the durability of mineral oil, for example, on human skin is enhanced by mixing the emollient with one of aminofunctional, amidofunctional, or carboxyfunctional, polysiloxanes.

5 This invention relates to a method of enhancing the durability of an emollient material on a substrate by forming a mixture of an emollient material and an effective amount of an organosilicon compound, and applying the mixture to the substrate to be treated. The organosilicon compound is either an aminofunctional, amidofunctional, or carboxyfunctional, polysiloxane.

Preferred embodiments of the present invention include an aminofunctional polysiloxane having the formula  $\text{Me}_3\text{SiO}(\text{Me}_2\text{SiO})_x(\text{MeRSiO})_y\text{SiMe}_3$  in which Me is methyl; R is the functional group  $-(\text{C}_4\text{H}_8)\text{NH}(\text{CH}_2)_2\text{NH}_2$ ; x is an integer from fifty to one thousand; and y is an integer from one to fifty preferably one to twenty. The amidofunctional polysiloxane preferably has the formula  $\text{Me}_3\text{SiO}(\text{Me}_2\text{SiO})_x(\text{MeRSiO})_y\text{SiMe}_3$  in which Me is methyl; R is the functional group  $-(\text{C}_4\text{H}_8)\text{NH}(\text{CH}_2)_2\text{NHCOCH}_3$ ; x is an integer from fifty to one thousand preferably fifty to one hundred; and y is an integer from one to fifty preferably one to ten. Preferably, the carboxyfunctional polysiloxane compound has the formula  $\text{Me}_3\text{SiO}(\text{Me}_2\text{SiO})_x(\text{MeRSiO})_y\text{SiMe}_3$  in which Me is methyl; R is the functional group  $-(\text{C}_3\text{H}_6)\text{COOH}$ ; x is an integer from fifty to one thousand preferably one hundred to three hundred; and y is an integer from one to fifty.

The emollient and the polysiloxane may be present in the mixture in the ratio of four to one, more preferably two to one. The mixture of the emollient and the polysiloxane may, if desired, include a volatile cyclic siloxane solvent, and the solvent can be a low viscosity polydimethylcyclosiloxane fluid, for example, which is a mixture of cyclic tetramers and pentamers and having a viscosity of about 2.5 centistokes measured at 25°C. The mixture should contain about five to ten percent by weight of the emollient and the polysiloxane in the volatile cyclic siloxane solvent. The emollient may be a material such as mineral oil, mink oil, lanolin oil, and petrolatum.

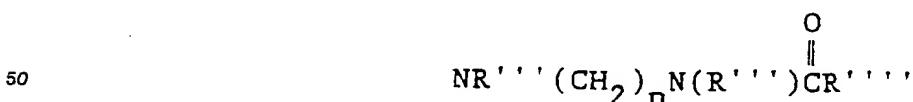
25 The invention also relates to a skin conditioning composition of enhanced durability which is a mixture of an emollient material and an effective amount of an organosilicon compound. The organosilicon compound, as noted above, is either an aminofunctional, amidofunctional, or carboxyfunctional, polysiloxane.

In accordance with the present invention, there is provided a concept which relates to the enhancement 30 of the durability of emollient materials with certain organosilicon compounds. The durability on human skin is enhanced by mixing an emollient with either one of an aminofunctional, amidofunctional, or carboxyfunctional, polysiloxanes. These aminofunctional, amidofunctional, and carboxyfunctional, organosilicon compounds are well known in the prior art, and such compounds as well as methods for preparing these 35 compounds can be found in U.S. Patent No. 4,477,514, which shows carboxyfunctional siloxanes; U.S. Patent No. 4,559,227, which shows aminofunctional siloxanes; and U.S. Patent No. 4,848,981, which shows the amidofunctional siloxanes, of the present invention. A brief description of these functional siloxanes is set forth hereinbelow.

The amine functional siloxane polymer has the formula

40  $\text{R}_{3-z}\text{Q}_z\text{SiO}[\text{R}_2'\text{SiO}]_x[\text{R}'\text{QSiO}]_y\text{SiQ}_z\text{R}_{3-z}'$

wherein R' denotes an alkyl group of 1 to 4 carbons or a phenyl group, with the proviso that at least 50 percent of the total R' groups are methyl; Q denotes an amine functional substituent of the formula  $-\text{R}''\text{Z}$ , wherein R'' is a divalent alkylene radical of 3 to 6 carbon atoms or a radical of the formulation 45  $-\text{CH}_2\text{CH}_2\text{CH}_2\text{OCH}_2\text{-CHOCH}_2-$  and Z is a monovalent radical selected from the group consisting of  $-\text{NR}_2'''$ ,  $-\text{NR}'''(\text{CH}_2)_n\text{NR}_2'''$ ; and



wherein R''' denotes hydrogen or an alkyl group of 1 to 4 carbons, R'''' denotes an alkyl group of 1 to 4 carbons and n is a positive integer from 2 to 6; z has a value of 0 or 1; x has an average value of 25 to 3000; y has an average value of 0 to 100 when z is 1, and y has an average value of 1 to 100 when z is 0.

The amidofunctional polysiloxane is a triorganosiloxane-endblocked polydiorganosiloxane having an average of 50 to 1000 siloxane units per molecule with an average of 1 to 50 of the siloxane unites per molecule being amide-containing siloxane units. The amide-containing siloxane units bear a substituent of

the formula



wherein n is 0 or 1, R' denotes an alkylene radical of 3 to 6 carbon atoms, and R'' denotes a hydrogen radical or an alkyl radical of 1 to 6 carbon atoms, X denotes an acyl radical of the formula

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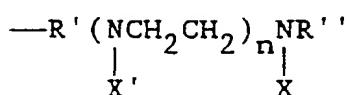


X' denotes a hydrogen radical or X, and R''' denotes an alkyl radical of 1 to 4 carbon atoms and substantially all other organic substituents in the polydiorganosiloxane being methyl groups.

The amidofunctional silicone component in accordance with this invention consists essentially of a triorganosiloxane-endblocked polydiorganosiloxane which contains amidoalkyl substituents. Triorganosiloxane-endblocked polydiorganosiloxanes (amidofunctional silicone) consist essentially of terminal triorganosiloxane units of the formula  $\text{R}_3\text{SiO}_{1/2}$  and backbone diorganosiloxane units of the formula  $\text{R}_2\text{SiO}_{2/2}$ . Trace amounts of other siloxane units in amidofunctional silicone, such as  $\text{SiO}_{4/2}$  and  $\text{RSiO}_{3/2}$ , which are normally present as impurities in commercial polydiorganosiloxanes may be present. Preferably there are no  $\text{SiO}_{4/2}$  units or  $\text{RSiO}_{3/2}$  units in the amidofunctional silicones.

The R radicals of the above siloxane units are substantially either amide-containing radicals of the formula

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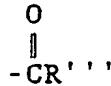
or methyl radicals. Minor amounts of other organic substituents which are normally present as impurities in commercial polydiorganosiloxanes may be present. It should be understood, for example, that the amidofunctional silicones of this invention are often prepared by acylation of corresponding aminofunctional silicones. Consequently, the amidofunctional silicones may also contain residual aminofunctional siloxane units. For example, siloxane unites such as  $\text{H}_2\text{NCH}_2\text{CH}_2\text{NHCH}_2\text{CH}(\text{CH}_3)\text{CH}_2\text{SiO}_{2/2}$  or  $\text{H}_2\text{NCH}_2\text{CH}_2\text{CH}_2\text{SiO}_{2/2}$  may also be present in the amidofunctional silicones useful in this invention. However, for the purposes of this invention it is preferred to employ silicone oils that do not contain significant levels (more than 25 percent of the number of amidofunctional substituents) of the unmodified aminofunctional siloxane units.

In the formula for the amide-containing radicals, R' denotes an alkylene radical of 3 to 6 carbon atoms, such as  $-\text{CH}_2\text{CH}_2\text{CH}_2-$ ,  $-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2-$ ,  $-\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}_2-$ ,  $-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2-$ , and  $-\text{CH}_2\text{CH}(\text{CH}_2\text{CH}_3)\text{CH}_2-$ . Amidofunctional silicones wherein the silicon bonded, amide-containing radicals have a trimethylene radical or an alkylated trimethylene radical, such as  $-\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}_2-$ , as the R' radical are preferred because of ease of synthesis and availability.

R'' denotes a hydrogen radical, which is a preferred R'' radical, or an alkyl radical of 1 to 6 carbon atoms, such as methyl, ethyl, propyl, butyl, and isobutyl.

In the formula for the amide-containing radicals, n has a value of 0 or 1, so that the radical may contain one or two nitrogen atoms. X denotes an acyl radical of the formula

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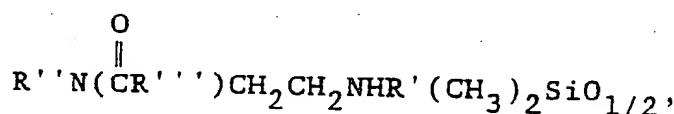


and X' denotes a hydrogen radical or X. In the acyl radical, R''' denotes an alkyl radical of 1 to 4 carbon atoms such as methyl, ethyl, propyl, isopropyl, or butyl.

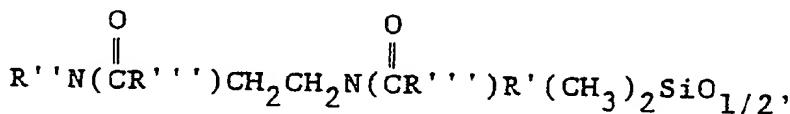
In accordance with the above, triorganosiloxane-endblocked polydiorganosiloxanes preferred for use in the method of this invention consists essentially of siloxane units selected from the following:

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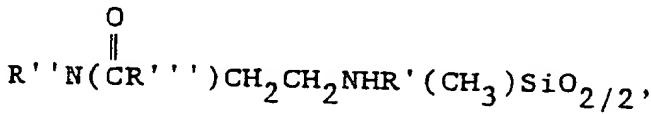
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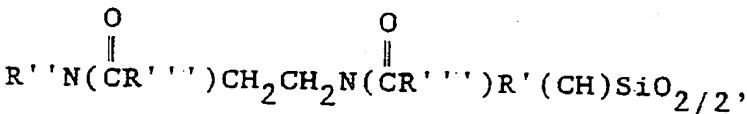
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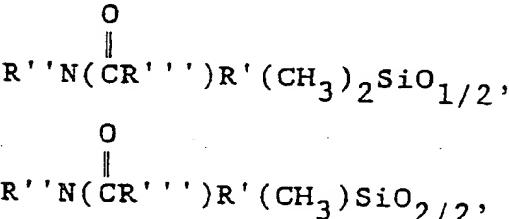
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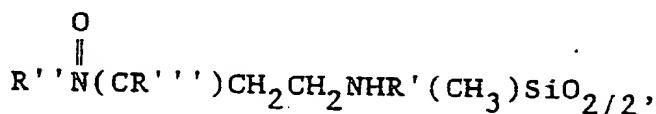
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(CH<sub>3</sub>)<sub>3</sub>SiO<sub>1/2</sub>, and (CH<sub>3</sub>)<sub>2</sub>SiO<sub>2/2</sub> where R', R'', and R''' have the same meanings as described above. It should be understood that any of the siloxane units having non-acylated nitrogen atoms can also be present in their salt form. It is well known that the salt form occurs when such polymers are neutralized by acids 35 such as mineral acids or carboxylic acids.

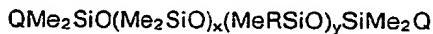
The silicone polymers of this invention may contain amide-containing siloxane units of the formula

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wherein R', R'', and R''' have the same meanings as described above. These amide-containing units have a 45 ratio of acyl groups to nitrogen atoms of about 0.5.

The carboxyfunctional silicones of the invention have the formula



50 wherein Me is a methyl radical, R is a carboxyfunctional radical, said carboxyfunctional radical being selected from the group consisting of carboxyalkyl radicals and carboxythioalkyl radicals, Q is selected from the group consisting of R, Me and OH groups, x has a value of 1 to 1000, and y has a value of 1 to 100.

As referred to herein, a carboxyfunctional radical is a monovalent radical which contains the —COOH radical, and is attached to a silicon atom of the main molecular chain by a divalent linking group. Direct 55 attachment to the silicon atom is through a silicon to carbon bond.

Divalent linking groups contemplated for use in the present invention are either alkylene groups containing from 2 to 9 carbon atoms, or thioalkylene groups, containing 2 to 8 carbon atoms and one sulfur atom present as a thioether group.

Those carboxyfunctional radicals wherein the divalent linking group is an alkylene group are referred to herein as carboxyalkyl radicals; those carboxyfunctional radicals wherein the divalent linking group is a thioalkylene group are referred to herein as carboxythioalkyl radicals.

Specific examples of carboxyalkyl radicals include, but are not limited to  $-\text{CH}_2\text{CH}_2\text{COOH}$ ,  $-\text{CH}_2\text{CH}(\text{CH}_3)\text{COOH}$ ,  $-\text{CH}_2\text{CH}(\text{C}_2\text{H}_5)\text{CH}_2\text{COOH}$ ,  $-\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}(\text{CH}_3)\text{CH}_2\text{COOH}$ , and the like. The  $-\text{CH}_2\text{CH}(\text{CH}_3)\text{COOH}$  radical is a preferred carboxyalkyl radical for the practice of the present invention.

Specific examples of carboxythioalkyl radicals include, but are not limited to  $-\text{CH}_2\text{CH}_2\text{SCOOH}$ ,  $-(\text{CH}_2)_3\text{SCOOH}$ ,  $-\text{CH}_2\text{CH}(\text{CH}_3)\text{SCH}_2\text{COOH}$ ,  $-\text{CH}_2\text{CH}_2\text{SCH}_2\text{COOH}$ ,  $-\text{CH}_2\text{CH}(\text{C}_2\text{H}_5)\text{SCH}_2\text{COOH}$ , and the like. The  $-\text{CH}_2\text{CH}_2\text{SCH}_2\text{COOH}$  radical is a preferred carboxythioalkyl radical for the practice of the present invention.

Examples of emollients and moisturizers which may be used in this invention include straight, branched or cyclic hydroxy compounds such as alcohols containing 1 to 30 carbon atoms; straight, branched, or cyclic carboxylic acids containing 1 to 31 carbon atoms; acid esters containing  $\text{C}_1$  to  $\text{C}_{30}$  carboxylic acids esterified with  $\text{C}_1$  to  $\text{C}_{30}$  alcohols; alcohol ethers containing 1 to 30 carbon atoms; alkanes of the formula  $\text{H}-(\text{CH}_2)_n-\text{H}$ , wherein  $n$  is 5 to 30; and siloxanes. Examples of such functional materials include 2-ethylhexyl oxystearate; arachidyl propionate; 2-ethylhexyl adipate; isopropyl myristate; ethanol; stearyl alcohol; propylene glycol; propionic acid; stearic acid; polyoxypropylene cetyl alcohol; polyoxypropylene lanolin alcohol; Carbowax® 300; petroleum jelly; mineral oil; aliphatic hydrocarbons such as mineral spirits; lanolin and lanolin derivatives such as acetylated lanolin and isopropyl lanolate; hexamethyldisiloxane; cyclic polydimethylsiloxane; linear polydimethylsiloxane; polypheylmethylsiloxane; and poly dimethyl(trimethyl)siloxane. Other phenyl, ethyl and vinyl substituted polysilanes may also be included in the products of this invention.

In order to illustrate the durability enhancement of the siloxanes of the present invention, durability enhancement data was collected for a variety of emollients. A soap washing test procedure was used in order to measure the influence of the silicones on the durability of the various emollients. The emollients considered were mineral oil, mink oil, lanolin oil, and petrolatum. The test procedure was used to measure silicone substantivity on human skin. Specifically, the method was based on Attenuated Total Reflectance/Fourier Transform Infrared Spectrophotometric (ATR/FTIR) analysis, in which prism skin studies were conducted and analyzed based on the reflection of energy at the interface. Instrumentation included a NICOLET Model 20DX FTIR system, and a HARRICK Scientific Skin Analyzer. The ATR studies involved contact of the skin sample and prism. A hydration procedure was employed in order to increase the softness and flexibility of the skin surface which resulted in a less variable contact between the skin and prism. This hydration procedure included placing a water soaked towel against the skin test site for one minute prior to actual spectra collection. A skin test site selected was an area of about eighty square centimeters, and about ten to twelve milligrams of each solution tested was applied to the skin test site area in the form of a thin film using a small paint brush. From the data collected, it was possible to calculate percentages of ingredients remaining on the skin following various soap wash sequences. The soap employed was a 0.5 weight percent solution of IVORY bar soap, and a soap rub is defined as two passes over the test area with the soap solution cupped in the palm of the hand. One soap wash procedure included fifteen soap rubs and ten rinse rubs under cool running tap water. The test site was the volar forearm. The test solutions were applied to the skin test site on the forearm in the form of a mixture of the various silicones and emollients, dissolved in a volatile silicone fluid of low viscosity, such as polydimethylcyclosiloxane which is a mixture of tetramer and pentamer having a viscosity of about 2.5 centistokes measured at 25 °C. The solution contained five to ten percent by weight of the mixture in the solvent. The solvent was allowed to evaporate from the volar forearm region for fifteen to thirty minutes prior to the institution of the measurement procedures. The site was hydrated as noted above and initial spectrum was collected. The data included tests conducted with and without the presence of the various silicones in the test mixture, and at least two test runs were conducted for each mixture.

A simplified test procedure is illustrated as follows. A test area on the forearm was marked, and the test area was washed with the soap solution using fifteen rubs, followed by rinsing with ten rubs under cool running water. Excess moisture was blotted from the forearm with a towel. After one minute, the skin was hydrated for one minute using a towel saturated with water which was held loosely over the test area. Excess moisture was blotted, and at the end of thirty seconds a background scan was run. The test mixture was applied to the skin test area and the solvent allowed to evaporate. The skin was again hydrated for one minute and excess moisture was blotted off. After thirty seconds, a scan was run of the test area which represented an Initial Condition. The test area was washed with the soap solution using fifteen rubs followed by ten rinses, and the excess moisture was blotted off. After one minute, the skin was hydrated for one minute, blotted, and at the end of thirty seconds, a scan was run of the test area which represented a First

Soap Wash Condition. Similar steps were repeated for second, third, and fourth, soap wash conditions. Baselines for infrared bands were defined and band heights were measured. The percent ingredient remaining on the skin was calculated using this data.

The following tables set forth the results of the foregoing procedures, and illustrate the concept of the present invention of enhancing the durability of various emollients with certain functional silicones. The tables indicate that the functional silicones enhance emollient durability and therefore provide a viable solution to dry, chapped, and rough skin, which results when the emollients are removed by washing. The siloxanes are soap wash resistant and have shown minimal or no dermal irritation. The carboxyfunctional siloxane is known to possess the least dermal irritation of the siloxane types tested. The functional silicones used in the tables conform to the formula  $\text{Me}_3\text{SiO}(\text{Me}_2\text{SiO})_x(\text{MeRSiO})_y\text{SiMe}_3$  in which Me is methyl and R is the functional group. Specifics of the R group and values of the integers x and y are set forth in the tables where appropriate. Unless otherwise indicated, the mixtures of emollient and silicone were in a ratio of four to one, and the mixtures were delivered in form of mixtures including a volatile cyclic siloxane. The compositions of the present invention may contain other adjuvants such as perfumes, fragrances, and preservatives, provided the addition of the adjuvant to the composition would not materially affect the basic and novel characteristics of the composition and would not materially change its fundamental characteristics.

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TABLE I
EMOLLIENTS - NO SILICONE  
PERCENT REMAINING

<u>Test Condition</u>	Mink Oil	Lanolin Oil	Mineral Oil	Petrolatum
Initial	100	100	100	100
1st wash	32	33	32	44
2nd wash	19	21	16	34
3rd wash	14	15	6	19
4th wash	---	---	4	16
5th wash	---	---	2	13

TABLE II
COMPOUNDS USED  
AMINE AND AMIDE FUNCTIONAL SILICONES

<u>Silicone</u>	<u>R-Group</u>	<u>M%R</u>	<u>X</u>	<u>Y</u>	<u>x/y</u>
A	iBuNH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	1	97	1	91/1
B	iBuNH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	2	96	2	48/1
C	iBuNH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	5	188	10	19/1
D	iBuNH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	0.7	296	2	148/1
E	iBuNH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	0.5	445.8	2.2	203/1
F	iBuNH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	1.7	440.4	7.6	58/1
G	iBuNH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	0.25	796	2	398/1
H	iBuNH(CH <sub>2</sub> ) <sub>2</sub> NHCOCH <sub>3</sub>	2	96	2	48/1

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TABLE III

PERCENT REMAINING  
MINERAL OIL AND SILICONES

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<u>Test Condition</u>	<u>SILICONES</u>							
	<u>H</u>	<u>A</u>	<u>B</u>	<u>C</u>	<u>D</u>	<u>E</u>	<u>F</u>	<u>G</u>
INITIAL	100	100	100	100	100	100	100	100
1st wash	46	56	53	47	46	46	50	52
2nd wash	28	38	35	35	40	32	34	41
3rd wash	22	32	28	30	32	32	28	32
4th wash	20	30	24	27	26	26	21	30
5th wash	17	25	20	24	22	25	20	19

TABLE IV

PERCENT REMAINING  
MINERAL OIL AND SILICONES

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<u>Test Condition</u>	<u>Polydimethylsiloxane (1)</u>			<u>H</u>	<u>Average A-G</u>
	<u>100DP</u>	<u>550DP</u>	<u>6800DP</u>		
INITIAL	100	100	100	100	100
1st wash	35	35	45	46	50
2nd wash	24	20	37	28	36
3rd wash	20	18	31	22	31
4th wash	18	16	27	20	26
5th wash	14	---	24	17	22

(1): 100 DP is 350 Centistoke fluid.

550 DP is hydroxy endblocked fluid with partial trimethyl capping.

6800 DP is a siloxane gum.

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TABLE V  
% Mineral Oil Remaining

5	<u>Sample Number</u>	<u>Polymer Description</u>	<u>Initial Condition</u>	<u>1st Wash</u>	<u>2nd Wash</u>	<u>3rd Wash</u>	<u>4th Wash</u>	<u>5th Wash</u>
10	1	No Silicone	100	32	16	6	4	2
15	2	H	100	56	34	25	20	14
20	3	A	100	40	26	18	16	14
25	4	B	100	52	27	18	19	13
30	5	C	100	40	28	25	18	14
35	6	D	100	58	36	30	17	8
40	7	E	100	66	41	34	26	24
45	8	F	100	58	37	28	17	15
50	9	G	100	33	30	18	20	5

TABLE VI

Compounds Used  
Carboxyfunctional Silicones

35	<u>Reference</u>	<u>R-Group</u>	<u>M%R</u>	<u>x</u>	<u>y</u>	<u>x/y</u>
40	J	iPr COOH	3	201.7	6.3	32/1
45	K	iPr COOH	0.7	296	2	148/1
50	L	iPr COOH	3.3	288	10	29/1
55	M	iPr COOH	15	253	45	6/1

TABLE VII

Percent Remaining  
Mineral Oil and Silicone

<u>Test Condition</u>	<u>J</u>	<u>K</u>	<u>L</u>
INITIAL	100	100	100
1st Wash	44	42	29
2nd Wash	30	33	19
3rd Wash	24	28	14
4th Wash	21	26	13
5th Wash	19	22	15

TABLE VIII

Mineral Oil and Silicone  
Percent Silicone Remaining

<u>Test Condition</u>	<u>Polydimethylsiloxane</u>			<u>H</u>	<u>A-G Amino Avg.</u>	<u>J-K Carboxylic Acid Avg.</u>
	<u>100DP</u>	<u>550DP</u>	<u>6800DP</u>			
INITIAL	100	100	100	100	100	100
1st Wash	35	35	45	46	50	43
2nd Wash	24	20	37	28	36	32
3rd Wash	20	18	31	22	31	26
4th Wash	18	16	27	20	26	24
5th Wash	14	--	24	17	22	20

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TABLE IX

**Mineral Oil and Silicone  
% Mineral Oil Remaining**

<u>Test Condition</u>	<u>No Sil.</u>	<u>J</u>	<u>K</u>	<u>L</u>
INITIAL	100	100	100	100
1st Wash	32	47	52	32
2nd Wash	16	30	36	20
3rd Wash	6	19	24	13
4th Wash	4	13	22	6
5th Wash	2	8	12	4

TABLE X

Two to One  
Mink Oil and Silicone

% Mink Oil Remaining

30	<u>Test Condition</u>	No Sil.	J	<u>Carboxylic Acid</u>	
				K	L
	INITIAL	100	100	100	100
	1st Wash	32	46	45	30
35	2nd Wash	19	28	28	17
	3rd Wash	14	26	25	10

The mixture of emollient and silicone can be delivered to the skin in the form of emulsions, microemulsions, solutions, dispersions, lotions, gels, aerosols, solid sticks, ointments, and creams.

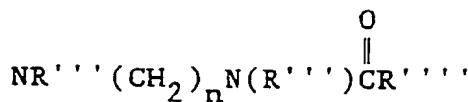
## Claims

1. A method of enhancing the durability of an emollient material on a substrate comprising forming a mixture of an emollient material and an effective amount of an organosilicon compound, and applying the mixture to the substrate to be treated, the organosilicon compound being selected from the groups consisting of (a) aminofunctional polysiloxanes having the formula

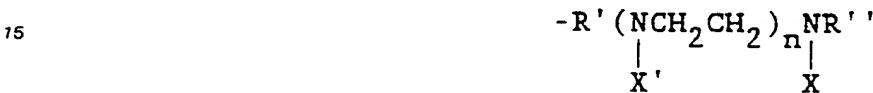


wherein R' denotes an alkyl group of 1 to 4 carbons or a phenyl group, with the proviso that at least 50 percent of the total R' groups are methyl; Q denotes an amine functional substituent of the formula

55 -R"Z, wherein R" is a divalent alkylene radical of 3 to 6 carbon atoms or a radical of the formula —CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OCH<sub>2</sub>-CHOHCH<sub>2</sub>— and Z is a monovalent radical selected from the group consisting of —NR<sub>2</sub>""", —NR"""(CH<sub>2</sub>)<sub>n</sub>NR<sub>2</sub>"""; and



wherein R''' denotes hydrogen or an alkyl group of 1 to 4 carbons, R'''' denotes an alkyl group of 1 to 4 carbons and n is a positive integer from 2 to 6; z has a value of 0 or 1; x has an average value of 25 to 3000; y has an average value of 0 to 100 when z is 1, and y has an average value of 1 to 100 when z is 0, (b) amidofunctional polysiloxanes having an average of 50 to 1000 siloxane units per molecule with an average of 1 to 50 of the siloxane units per molecule being amide-containing siloxane units bearing a substituent of the formula



20 wherein n is 0 or 1, R' denotes an alkylene radical of 3 to 6 carbon atoms, and R'' denotes a hydrogen radical or an alkyl radical of 1 to 6 carbon atoms, X denotes an acyl radical of the formula

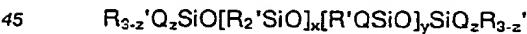


X' denotes a hydrogen radical or X, and R''' denotes an alkyl radical of 1 to 4 carbon atoms and substantially all other organic substituents in the polysiloxane being methyl groups, and (c) carboxyfunctional polysiloxanes having the formula

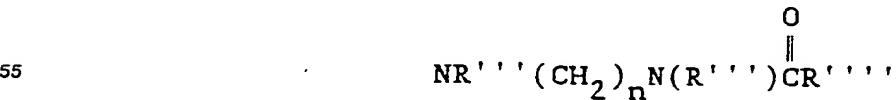


wherein Me is a methyl radical, R is a carboxyfunctional radical, said carboxyfunctional radical being selected from the group consisting of carboxyalkyl radicals and carboxythioalkyl radicals, Q is selected from the group consisting of R, Me and OH groups, x has a value of 1 to 1000, and y has a value of 1 to 100.

2. The method of claim 1 in which the mixture of the emollient and the polysiloxane includes a volatile cyclic siloxane solvent.
- 40 3. A skin conditioning composition of enhanced durability consisting essentially of a mixture of an emollient material and an effective amount of an organosilicon compound, the organosilicon compound being selected from the group consisting of (a) aminofunctional polysiloxanes having the formula



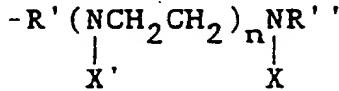
wherein R' denotes an alkyl group of 1 to 4 carbons or a phenyl group, with the proviso that at least 50 percent of the total R' groups are methyl; Q denotes an amine functional substituent of the formula —R''Z, wherein R'' is a divalent alkylene radical of 3 to 6 carbon atoms or a radical of the formula —CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OCH<sub>2</sub>—CHOHCH<sub>2</sub>— and Z is a monovalent radical selected from the group consisting of —NR<sub>2</sub>''', —NR'''(CH<sub>2</sub>)<sub>n</sub>NR<sub>2</sub>'''; and



wherein R''' denotes hydrogen or an alkyl group of 1 to 4 carbons, R'''' denotes an alkyl group of 1 to 4

carbons and n is a positive integer from 2 to 6; z has a value of 0 or 1; x has an average value of 25 to 3000; y has an average value of 0 to 100 when z is 1, and y has an average value of 1 to 100 when z is 0, (b) amidofunctional polysiloxanes having an average of 50 to 1000 siloxane units per molecule with an average of 1 to 50 of the siloxane units per molecule being amide-containing siloxane units bearing

5 a substituent of the formula



wherein n is 0 or 1, R' denotes an alkylene radical of 3 to 6 carbon atoms, and R'' denotes a hydrogen radical or an alkyl radical of 1 to 6 carbon atoms, X denotes an acyl radical of the formula

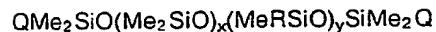
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X' denotes a hydrogen radical or X, and R''' denotes an alkyl radical of 1 to 4 carbon atoms and substantially all other organic substituents in the polysiloxane being methyl groups, and (c) carboxyfunctional polysiloxanes having the formula

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wherein Me is a methyl radical, R is a carboxyfunctional radical, said carboxyfunctional radical being selected from the group consisting of carboxyalkyl radicals and carboxythioalkyl radicals, Q is selected from the group consisting of R, Me and OH groups, x has a value of 1 to 1000, and y has a value of 1 to 100.

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4. The composition of claim 3 in which the mixture of the emollient and the polysiloxane includes a volatile cyclic siloxane solvent.

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EUROPEAN SEARCH  
REPORT

EP 91 10 3231

DOCUMENTS CONSIDERED TO BE RELEVANT					
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)		
X	EP-A-0 078 597 (TORAY SILICONE CO., LTD) * Page 6, paragraph 4 - page 11, paragraph 3, claims * - - -	1-4	A 61 K 7/48 A 61 K 7/06		
X	WO-A-8 904 161 (THE GILLETTE CO.) * Page 2, line 22 - page 9, line 28; examples; claims * - - -	1,3			
D,X	US-A-4 559 227 (CHANDRA et al.) * Column 3, line 65 - column 8, line 4; column 9, line 30 - column 11, line 13 * - - -	1,3			
X	EP-A-0 095 238 (DOW CORNING CORP.) * Page 2, line 30 - page 7, line 11 * - - -	1,3			
X	EP-A-0 174 097 (DOW CORNING CORP.) * Page 9, line 18 - page 10, line 23; page 12, lines 9-31; examples; claims * - - -	1,3			
X	US-A-4 501 619 (GEE) * Column 2, line 45 - column 10, line 16, claims * - - - -	1,3			
			TECHNICAL FIELDS SEARCHED (Int. Cl.5)		
			A 61 K		
The present search report has been drawn up for all claims					
Place of search	Date of completion of search	Examiner			
The Hague	10 June 91	COUCKUYT P.J.R.			
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